

NEW TECHNOLOGY FOR EMISSION CONTROL IN MODERN KRAFT RECOVERY BOILERS

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Abstract

The recovery boiler in an old pulp mill is often a large source of SO_2 , TRS, NO_x and CO emissions to the environment.

The emissions from modern kraft recovery boilers are much lower than from 20– 30 year old units. It is the state of the art to rebuild old units to eliminate SO_2 , TRS and CO emissions and at the same time increase the capacity of the unit. In the last couple of years, techniques have also been demonstrated that minimize the NO_x emissions.

Introduction

The recovery boiler is one of the most important production units of the kraft pulp mill. Being both a chemical reactor for recovering of chemicals from the pulping process as well as an energy recovery process for steam and power production, the recovery boiler constitutes the heart of the kraft pulp mill.

The first Tomlinson recovery boiler in Europe was delivered by Götaverken to the Husum kraft pulp mill in 1936. For more than fifty years the Tomlinson type recovery boiler has been continuously developed to meet increasing demands for capacity, thermal efficiency, safety, availability as well as emission control.

The demands for emission control were for many years not very high and recovery boilers emitted large amounts of particles, SO_2 and TRS (total reduced sulfur compounds). Recovery boilers equipped with direct contact evaporators emitted for example well above 100 ppm TRS to the atmosphere.

During the last decades, great improvements have, however, been made regarding control of emissions from the recovery boilers as well as the rest of the pulp mill.

The emission of sulfur in Sweden is illustrated in Table 1 for the years 1980 and 1988. The total sulfur emission in Sweden has been reduced by about 60 % during 1980's. During the same time, the Pulp & Paper Industry reduced its emission of sulfur by more than 70 %. The specific emission in 1988 was 1.7 kg S/t pulp.

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Year		1980	1988	
Total	t S/year	257 000	106 000	
Pulp & Paper	t S/year	48 000*	14 000	
Kraft Pulp Mills	t S/year	32 800*	10 000	
Kraft Pulp Mills	kg S/t pulp	6.1	1.7	
* 50 – 55 % from oil firing				

Less than 15% of the total emission of sulfur in Sweden now comes from the Pulp & Paper Industry, the dominating source being the energy generation sector. The emission of sulfur originating both from domestic sources and from surrounding countries is, however, still too high to avoid acidification. Although the Pulp & Paper Industry only contributes with a minor part of the emission of sulfur, a single pulp mill still constitutes a large single source for sulfur emission. further reduction in permitted sulfur emission has been announced by the Swedish authorities both for energy generation and for pulp & paper production.

During the last decade an increasing knowledge with respect to the negative environmental effect of nitrogen oxides ($NO_x = NO + NO_2$) through acidification, overfertilization and ozone formation has resulted in more stringent demands for limitation of NO_x emission from all combustion processes.

The concentration of NO_x in the flue gases from recovery boilers is in the range of 4070 mg/MJ (based on net heating value of the fuel in reducing atmosphere) which, however, is low in comparison with other combustion processes. Despite this relatively low flue gas NO_x concentration, kraft recovery boilers still are great NO_x emission sources because of their totally large flue gas volumes.

Typical average values for the concentration of the most important pollutants in the recovery boiler flue gas in a Swedish kraft pulp mill are shown in Table 2.

Particulates	mg/m ³ n	200
TRS	ppmv	0 - 5
SO ₂	ppmv	50 - 300
NO _x	ppmv	60 - 100*
со	ppmv	300
Dry gas at 3% O ₂		
* 40 – 70 mg NO ₂ /m ³ n		

TABLE 2. Typical emissions from Swedish kraft recovery boilers

In Table 3 are shown both guaranteed and performance test data from a recently started up North-American Götaverken recovery boiler. These values represent the most stringent emission limits of all delivered Götaverken recovery boilers to date.

		Guarantees	Performance test
Particulates	mg/m ³ n	70	40
TRS	ppmv	7	0.1
SO ₂	ppmv	210	42*
NO _x	ppmv	110	98
со	ppmv	300	280
VOC	mg/m ³ n	65	7
Dry gas at 3% O ₂			
VOC = Volatile Orga	nic Compou	nds	
* S/Na ₂ = 28 - 30			

TABLE 3.	Emissions from a modern North-American Götaverken kraft recovery
	boiler

Below is discussed the status of recovery boiler emission control with respect to internal measures for reduction of TRS, SO_2 and CO as well as new developments for reductions of NO_x .

Status of Recovery Boiler Emission Control

The fundamental mechanism for the formation and control of SO_2 and TRS in recovery boilers have been known for many years and with a gradual introduction of modern combustion air system, computer control systems, automatic air port cleaning equipment as well as high black liquor dry solids evaporators it has been possible to obtain the low emission levels given in Tables 2 and 3.

In Figure 1 is illustrated the primary emission of sodium and sulfur as well as the resulting SO_2 emission versus the furnace temperature based on a thermodynamic calculation presented by Borg et al (1). According to the model, SO_2 will be formed if more sulfur is emitted from the combustion zone than the corresponding emission of sodium for sulfur capturing as Na_2SO_4 . The figure shows a very high influence of the furnace temperature on the SO_2 emission.

For optimum control of combustion conditions in the recovery boiler, Götaverken in 1960 introduced the separately fed three level air system. With the strong relationship between SO_2 and combustion temperature, the proper adjustment of combustion air to the different air levels is very important.

Equally important for a proper combustion and low SO_2 emission is the proper distribution of air around the boiler on all air levels. For this reason, Götaverken in 1973 introduced the RODDINGMASTER System for automatic cleaning of primary air ports from frozen smelt at present frequent intervals. If the frozen smelt is not constantly removed, air flow will vary causing upsets in reduction efficiency and emissions. With RODDINGMASTER the combustion process will stabilize which makes computer control possible.

Today, the RODDINGMASTER system has also been installed in the secondary air register in some recovery boilers for minimum air flow variation in the combustion zone.

In 1981, the Mönsterås Pulp Mill in Sweden initiated a joint development project together with ASEA, Götaverken Energy and STFI for better recovery boiler control. The primary goal for the project was better SO_2 emission control an increased availability and liquor throughput. The project involved fundamental laboratory studies, development of new control methods and new equipment for measurement and control.

Mönsterås Mill began operating with the new Advanced Control system (BLRBOMASTER) in May 1984. Already during its first year of operation, Mönsterås experienced considerable savings in operational costs, reduced excess air by more than 50%, reduced SO₂ emission to 2.3 kg SO₂/t pulp and practical elimination of TRS.

The BLRBOMASTER System consists of nine control modules which are illustrated in Figure 2.

The burning intensity module applies the fundamental relationship illustrated in Figure 1 between burning intensity (furnace temperature) and sodium/sulfur emission from the furnace. Changes in the sodium/sulfur emission from the furnace can be determined from the dust content as well as SO_2 content of the flue gas and the Ph value of the flue gas dust. This data is utilized in computing the burning intensity coefficient of the boiler.

The purpose of the control is to stabilize the burning intensity, which involves a qualified trend analysis because of the marked nonlinearity between for example SO_2 and the quantities of primary and secondary air respectively as illustrated in Figure 3 as presented by Herngren (2).

Final combustion control is achieved by measurement of the flue gas content of CO, which is used as an indicator of uncombusted substances, and O_2 followed by control the tertiary air flow and pressure to the boiler. This control provides stable operation of excess air at varying production rates while retaining a suitable CO content in the flue gas.

During the 80's the black liquor dry solids in new evaporation plants have increased from 68% up to 75% DS. The dry solids standard of the 70's was 65% and many pulp mills operate their evaporation plant (from the 70's) at this level.

With an increased black liquor dry solids to our most recent recovery boiler installations, the combustion has stabilized and the burning intensity has increased. As a result of this increase in burning intensity, the emission of SO_2 has practically been reduced to zero.

The added effect of the following discussed measures enables very low emission of both SO_2 and TRS as well as CO over a wide load range and up to high sulfidity levels.

Modern three level air system RODDINGMASTER for automatic air port cleaning BLRBOMASTER for computer control High dry solids evaporation

The normal operation range for SO_2 emission versus S/Na_2 in the black liquor is illustrated in Figure 4 for a black liquor dry solids content of 65–67%. The lower level of the operation range represents the SO_2 flue gas concentration with modern three level air system, RODDINGMASTER and BLRBOMASTER in operation. With increased dry solids, the curve will move to higher S/Na_2 values. It is obvious from the figure that a proper control of SO_2 emissions to low levels requires good control of the sulfur balance for the whole mill.

With the BLRBOMASTER system in operation in the referred North-American recovery boiler in Table 3 the best balance for both reduction efficiency, combustion stability, and NO_x emission was achieved when SO_2 was balanced at 30–50 ppm.

Pulp mills with high intakes of chlorides to a too high burning intensity in the recovery furnace might make chlorides accumulate in an internal recovery boiler recycle stream of NaCl with a subsequent risk of increased fouling of the heating surfaces.

Development of NO, Reduction Techniques

With the development of new techniques for recovery boiler control the trend for particulates, SO_2 , TRS and CO emissions have been decreasing.

The formation mechanisms for the emission components SO_2 , NO_x , CO, and TRS are interrelated in the recovery furnace.

The interrelationship between CO and TRS is fairly easy to understand since both components can be seen as a result of incomplete combustion. There is an interrelation between NO_x and CO as well as between NO_x and SO_2 which makes it difficult to achieve low emission figures for all three pollutants at the same time.

This interrelation has also caused a trend of slightly increasing NO_x emissions in modern high efficiency boilers. Some of the reasons for this trend are:

Increased black liquor dry solids Increased furnace loads High degree of SO₂ reduction Demands for lower CO/TRS emission

The strongest interrelation is between NO_x and CO since the formation of NO_x and combustion of CO both depend upon the concentration of O₂ in the combustion zone. For the North-American boiler referred to in Table 3, this interrelation necessitated a special tuning of the BLRBOMASTER System delivered with the recovery boiler to be able to meet both NO_x and CO guarantees simultaneously.

To be able to meet the expected more stringent demands both for NO_x and CO emissions from recovery boilers, Götaverken Energy has investigated both selective non-catalytic reduction (SNCR) as well as combustion control measures. A comparison of relative costs and efficiencies of both combustion control and SNCR as well as selective catalytic reduction (SCR) presented by Epperly et al (3) is shown in Figure 5.

The SNCR test was initiated by Södra Skogsägarna who commissioned a full scale demonstration of the $NO_xOUT(R)$ Process in the existing Götaverken recovery boiler at Mönsterås in co-operation with Petrokraft.

In the NO_xOUT process NO is reduced by urea to nitrogen, carbon dioxide and water according to the reaction:

 $2NO + (NH_2)2CO + 1/2O_2 \rightarrow 2N_2 + CO_2 + 2H2O$

The reaction normally occurs within a narrow temperature range around 1000° C. If the temperature is too high, more NO_x is produced and if the temperature window for reduction is widened and the by-production of ammonia is suppressed by chemical enhancers.

Figure 6 shows the NO_x emission at the beginning of the final two weeks continuous verification operation in June–July 1990.

The by-product of NH_3 was only 5 mg/m³n dry gas with the NO_xOUT process in operation.

The result of the demonstration showed that more than 60% reduction was achieved despite a low initial NO_x level. No disturbances or other negative effects on the operation of the recovery boilers or in the chemical recovery cycle were observed during the test.

In addition to the SNCR demonstration in Mönsterås, Götaverken Energy has improved the control of the combustion process by development of the air system of the recovery boiler. Table 4 shows the result from a test with this modified air system in a recovery boiler. The results show a potential NO_x reduction of at least 30–40% in spite of low initial NO_x level.

TABLE 4. I	NO, reduction	with modified	tertiary air system
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Test results from Mönsterås Oct-Nov 1990				
Air system	NO _x ppm	NO _x % reduction		
Normal	50	0		
Modified	30	40		
5-10% tertiary air flow of	total			

For both these methods, the resulting NO_x concentration is less dependent on the amount of excess air which enables us to reduce CO (together with other minor resulting combustible compounds) with O_2 control as well as air control without increasing the NO_x emission.

NO_x-Reduction through Combustion Control Measures

The low NO_x emissions from the recovery boiler has been attributed by, among others, Anderson and Jackson (4) to the low nitrogen content and the low heat value as well as the normal performance of the boiler with air supply on several levels. The same authors also suggest that further NO_x reduction may be achieved through manipulation of air distribution and the characteristics and distribution of the fuel.

This kind of test has been carried out, within the scope of the SSVL-project Miljö 90 (5), by way of air manipulation for both temperature and distribution. None or an insignificant effect on the NO_x emission of the recovery boiler has been reached without disturbing other operation parameters.

These results also correspond with Götaverken's experiences which have shown that it is possible to have a certain amount of NO_x control by manipulation of a modern equipped recovery boiler. Unfortunately, the degree of manipulation is relatively limited.

To investigate if it is possible to reach more drastic effects on the NO_x reduction through so called combustion control measures in a recovery boiler, a Swedish recovery boiler was equipped with some experimental air registers on different levels in the upper furnace.

The possibilities with different combinations of so called overfire air systems to reduce the NO_x emission were tested on two occasions during October-November 1990.

The results were very positive and indicated a reduction of 30–40% with a maximum performance design already at 6–8% overfire air. Furthermore, it is likely that additional improvements can be reached if a larger amount of overfire air is being used.

The detailed results have, to some extent, been confirmed in one of Götaverken's North-American recovery boilers where a very high dry solids content has allowed a high proportion of the air provided through a simpler overfire air system.

The potential for reduction of the NO_x emission of the recovery boiler with an overfire air system and with NO_xOUT process respectively is indicated in the table below.

	Reduction %	NO _x mg/MJ
Standard System	0	60
Overfire air system	30-50	30-40
NO _x OUT	60-70	20

TABLE 5. Potential for NO_x reduction in recovery boilers

Götaverken has, through these test, participated in developing two techniques, which separately or in combination, considerably reduce the NO_x emission of the recovery boiler to levels far below those which are normally reached in alternative combustion processes.

Conclusions

Compared to the energy generation sector, the emissions from kraft pulp mills are generally low. However, the recovery boiler in a pulp mill constitutes a large single source of emission because of its totally large flue gas volumes.

It has been shown that the technical potential for reduction of the different pollutants is still very high, although all the lowest limits might not be economically justified in all areas today. Table 6 shows a summary of estimated technical and econimical emission target ranges for the different pollutants from a recovery boiler.

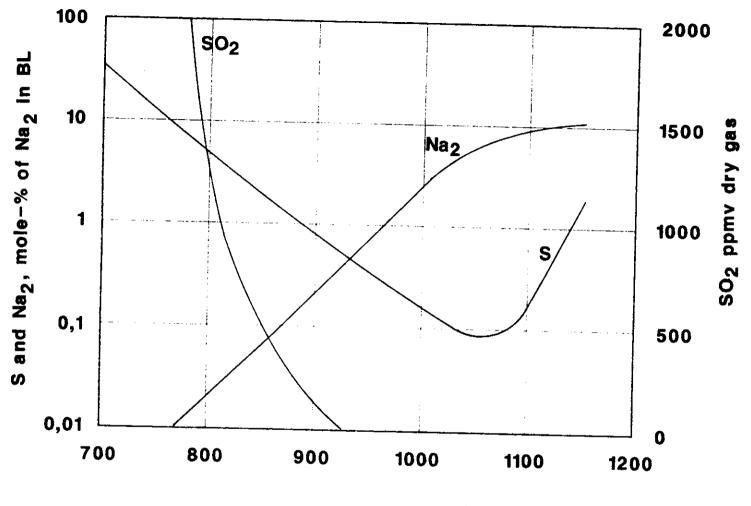
Verified results of various measures for reduction of the emissions of particles, SO_2 , TRS, NO_x and CO show that the recovery boiler has a potential to become one of the most environmentally friendly combustion processes.

Pollutant		Typical	Future	Control measures
Particulates	mg/m ³ n	200	50 - 100	
TRS	ppmv	0 - 5	0 - 2	Related to CO
SO2	ppmv	50 - 300	0 – 50	Modern air system RODDINGMASTER BLRBOMASTER Target in Sweden is 0.8 kg S ptp
NO _x	ррту	60 - 100	20 - 40	NO _x OUT(R) Process or modified air system. Measures up to 40 SEK/kg NO _x is proposed in Sweden
со	ppmv	300	100 - 300	BLRBOMASTER NO _x control
Dry gas at 3 % O ₂				

TABLE 6.	Summary of emission control potentials in modern kraft recovery	
	boilers	

References

- 1. Borg A., Tder A., and Warnqvist B.; TAPPI 57(1974), 1, 126–129
- 2. Epperly W., Broderick R., Peter-Hoblyn J.; Annual meeting of the American Power Conference April 20, 1988
- 3. Herngren T.; Svensk Papperstidning Nr. 15, 1989, 18-22
- 4. Anderson Peter H., Jackson James.; TAPPI J., Jan. 1992, 115 – 118
- 5. Stripple Håkan, Cooper David, Boström Curt-Åke; Miljö 90, Rapport nr 79



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Furnace temperature, °C

Figure 1. Emission of S, Na₂ and SO₂ versus furnace temperature $S/Na_2 = 30$ mole-%

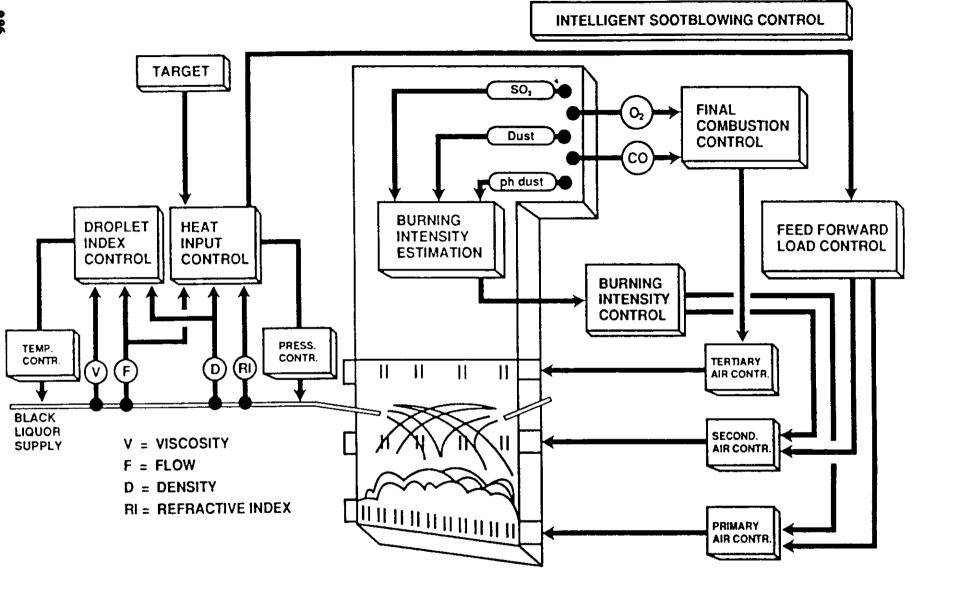


Figure 2. BLRBOMaster automated control package

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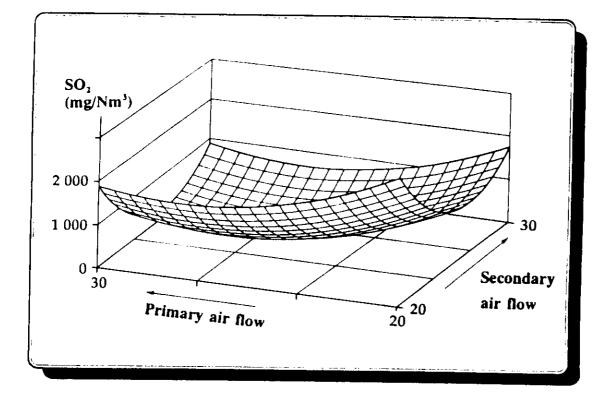


Figure 3. Emission of SO₂ versus primary and secondary air flow

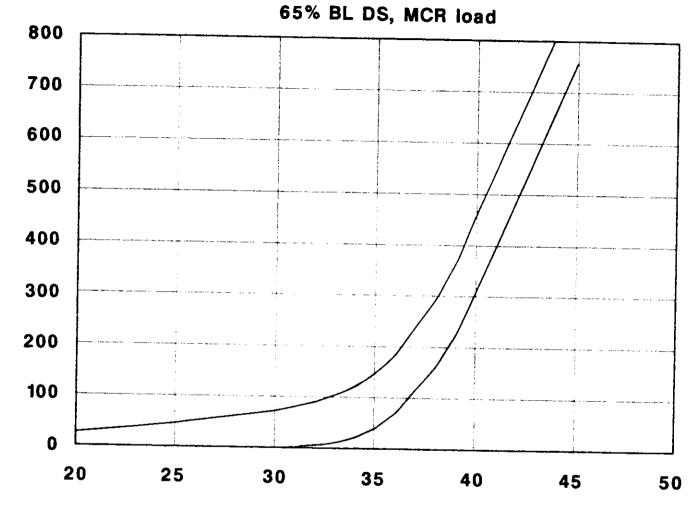
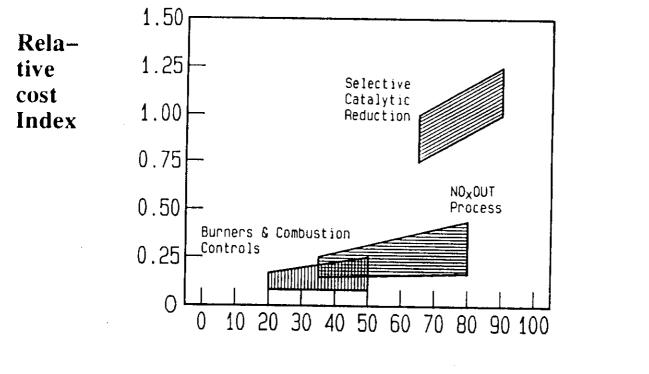




Figure 4. Emission of SO₂ versus S/Na₂ ratio operating range for 65-67% DS, 100% MCR loa

SO2, ppmv dry gas



% NO_x Emission Reduction

Figure 5. NO_x emission control technologies Comparison of relative cost and effectiveness

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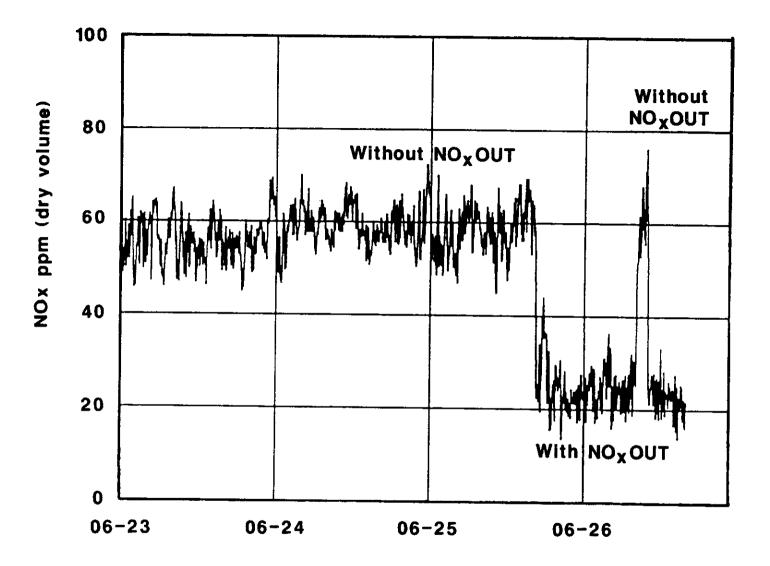


Figure 6. NO_x-emissions from the period of start-up of the continuous run.

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